

Easily accessible ferrocenyl N-P/S type ligands and their applications in asymmetric allylic substitutions

Fuk Loi Lam,^a Terry T. L. Au-Yeung,^a Hong Yee Cheung,^a Stanton H. L. Kok,^a Wing Sze Lam,^a Kwok Yin Wong^a and Albert S. C. Chan^{a,b,*}

^aOpen Laboratory of Chirotechnology of the Institute of Molecular Technology for Drug Discovery and Synthesis, Department of Applied Biology and Chemical Technology, The Hong Kong Polytechnic University, Hung Hom, Hong Kong

^bState Key Laboratory of Chinese Medicine and Molecular Pharmacology, Shenzhen, China

Received 6 December 2005; accepted 14 December 2005

Available online 10 March 2006

Abstract—Easily accessible novel 1,2-disubstituted phosphinamidite-thioether ligands based on a ferrocene motif have been developed, and successfully applied for asymmetric allylic substitutions with excellent yields and enantioselectivities.
© 2006 Elsevier Ltd. All rights reserved.

1. Introduction

The transition metal-catalyzed asymmetric allylic substitution has become a powerful tool for enantioselective carbon–carbon and carbon–heteroatom bond formation.¹ Several classes of chiral ligands, such as bisphosphines,¹ monodentate phosphines,² and P/N mixed-donor ligands^{3–5} have been extensively studied and proven to be effective ligands for Pd-catalyzed asymmetric allylic substitution reactions. In the literature, few reports concern the use of chiral P/S mixed donors for metal-catalyzed asymmetric reactions^{6–9} (Fig. 1). Seminal work by Evans et al. showed that O-P/S mixed-donor ligands could mediate Rh-catalyzed hydrogenation reactions and Pd-catalyzed allylic alkylation with enantioselectivity up to 98% ee.⁷ Recent works by Carretero et al. also revealed that several P/S ligands were effective for Pd-catalyzed allylic substitution reactions,⁸ ring opening of oxa- and aza-bicyclic alkenes,^{9,12} aza Diels–Alder reactions,¹⁰ and 1,3-dipolar cycloaddition of azomethine ylides.¹¹

Ferrocene-based chiral phosphines have found important applications for metal-catalyzed asymmetric reactions.¹³ Recently, we developed chelating ferrocenyl phosphine-phosphinites **5**, phosphine-phosphoramidites **6**, and phosphine-phosphites **7**. These have been success-

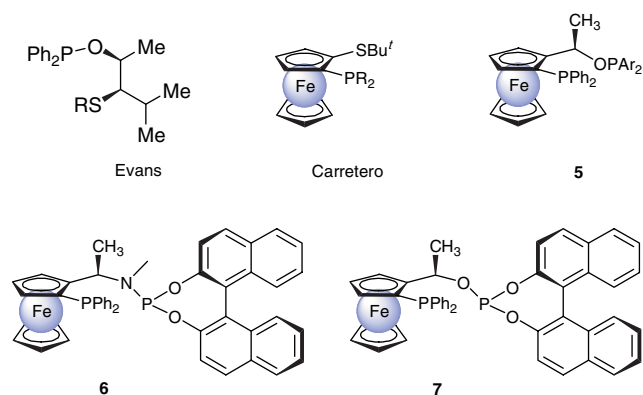


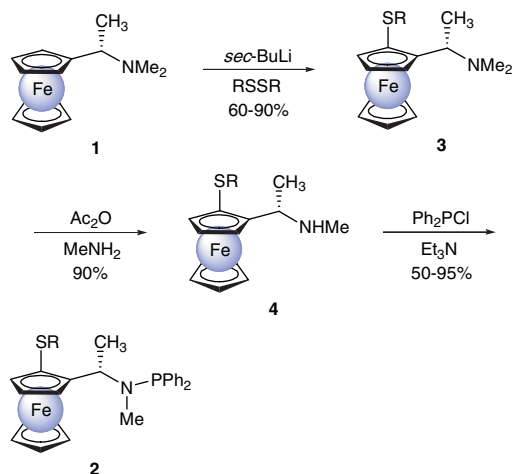
Figure 1.

fully applied for Rh-catalyzed hydrogenation of dehydro- α -amino acid derivatives to give hydrogenated product with excellent enantiopurity.¹⁴ Further to these efforts, we herein report that bidentate ferrocenyl phosphinamidite-thioether ligands, such as **2**, are promising ligands for asymmetric catalysis. The results of their application in asymmetric allylic alkylation and amination will be described.

2. Results and discussion

Scheme 1 depicts the synthetic route for ferrocenyl phosphinamidite-thioether ligands **2**. Starting from the

* Corresponding author. Tel.: +852 27665607; fax: +852 23649932; e-mail: bcachan@polyu.edu.hk



Scheme 1. Synthesis of the ferrocenyl N-P/S ligands **2**.

commercially available chiral Ugi's amine,¹⁵ diastereoselective *ortho*-lithiation using *sec*-BuLi/Et₂O, followed by quenching with disulfides (R = Et, ^tBu, Ph) afforded the 1,2-disubstituted ferrocenyl amines **3a–c** in 60–90% yields.¹⁶ Treatment of **3** with Ac₂O and methylamine furnished the ferrocenyl methylamine **4** in >90% yield. The ferrocenylamine was then converted to the phosphinamidite **2** (50–95% yield) by phosphinylation using Et₃N and Ph₂PCl.¹⁷

When 1,3-diphenyl-2-propenyl acetate (0.5 M solution) was treated with dimethyl malonate (3 equiv) in toluene containing LiOAc as an additive (2 mol %), BSA (3 equiv), [Pd(η³-C₃H₅)Cl]₂ (2 mol %) and (*S,R_p*)-FerroNPS-Et **2a** (4.2 mol %) at room temperature, the alkylated product was produced in >99% conversion and 87.9% ee based on chiral HPLC analysis (Table 1, entry 1). For the asymmetric substitution of 1,3-diphenyl-2-propenyl acetate, using NaOAc or KOAc as an additive

Table 1. Pd-catalyzed asymmetric allylic alkylation using (*S,R_p*)-FerroNPS-Et **2a** as a chiral ligand with an addition of metal acetate as an additive in various solvents^a

Entry	Additive	Solvent	Time (h)	Conv. (%) ^b	ee (%) ^c
1	LiOAc	THF	12	>99	87.9 (<i>R</i>)
2	NaOAc			>99	86.6 (<i>R</i>)
3	KOAc			>99	86 (<i>R</i>)
4	Zn(OAc) ₂			>99	90.2 (<i>R</i>)
5		CH ₂ Cl ₂		>99	88.6 (<i>R</i>)
6		CH ₃ CN		>99	86.1 (<i>R</i>)
7		Toluene		>99	91.8 (<i>R</i>)

^a Reaction conditions: [Pd(η³-C₃H₅)Cl]₂ (2 mol %), ligand **2** (4.2 mol %), dimethyl malonate (3.0 equiv), BSA (3.0 equiv), additive (2.0 mol %), and 0.5 M of concentration, room temperature.

^b The conversion was determined by ¹H NMR analysis of the crude reaction mixture.

^c The % ee value was determined by HPLC on a Chiralpak AD column (1.0 mL/min, *n*-Hex/ⁱPrOH = 95:5).

did not result in better enantioselectivity (ca. 86% ee; entries 2 and 3). Herein, Zn(OAc)₂ was found to be the best additive; up to 90.2% ee was attained for the allylic substitution reaction (entry 4).

The effect of solvent for the Pd-**2a** catalyzed allylic substitution reaction of 1,3-diphenyl-2-propenyl acetate was also investigated. As shown in Table 1, CH₂Cl₂ and CH₃CN are effective solvents for the allylic substitution reaction, and enantioselectivities of 88.6% and 86.1% ee were observed, respectively (entries 5 and 6). The best result (91.8% ee) was achieved when toluene was employed as solvent (entry 7).

Under the optimized reaction conditions: [Pd(η³-C₃H₅)Cl]₂ (2 mol %); **2** (4.2 mol %), dimethyl malonate (3.0 equiv), BSA (3.0 equiv), and Zn(OAc)₂ (2.0 mol %) in toluene at room temperature, the effectiveness of other ferrocenyl phosphinamidites was tested using 1,3-diphenyl-2-propenyl acetate as substrate. Our results in Table 2 show that **2b** and **2c** bearing bulky R groups (^tBu and Ph) are effective ligands for the Pd-catalyzed allylic substitution using diethyl malonate with ca. 93% ee being attained. Apparently, the bulkier thioether groups would necessitate longer reaction time (up to 120 min) albeit with improved enantioselectivities.

Table 2. A study of influence of the thioether group of (*S,R_p*)-FerroNPS ligands **2** on Pd-catalyzed AAA reaction^a

Entry	Ligand	Concn (M)	Time (min)	Yield (%) ^b	ee (%) ^c
1	2a (R = Et)	0.5	45	97	91.8 (<i>R</i>)
2	2b (R = ^t Bu)		90	94	92.7 (<i>R</i>)
3	2c (R = Ph)		120	94	93.5 (<i>R</i>)

^a Reaction conditions: [Pd(η³-C₃H₅)Cl]₂ (2 mol %), ligand **2** (4.2 mol %), dimethyl malonate (3.0 equiv), BSA (3.0 equiv), Zn(OAc)₂ (2.0 mol %), and toluene as solvent at room temperature.

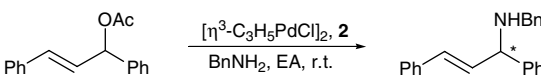
^b Isolated yield.

^c The % ee value was determined by HPLC on a Chiralpak AD (1.0 mL/min, *n*-Hex/ⁱPrOH = 95:5).

Having achieved enantioselective C–C bond formation using the Pd-**2a** catalyzed allylic substitution reaction, we also evaluated the ferrocenyl phosphinamidite ligands for analogous C–N bond formation. Treatment of 1,3-diphenyl-2-propenyl acetate (0.25 M) with benzylamine (3 equiv) in ethyl acetate containing [Pd(η³-C₃H₅)Cl]₂ (2 mol %) and (*S,R_p*)-FerroNPS-Et **2a** (4.2 mol %) at room temperature, afforded the product allyl amine in 98% yield and 89.1% ee. Similarly, other ferrocenyl ligand derivatives **2b** and **2c** were also found to effect the allylic amine substitutions with enantioselectivities of 91.5% and 81.7% ee, respectively (Table 3).

3. Conclusion

In conclusion, we have successfully developed a new class of easily accessible ferrocene-based 1,2-disubstituted phosphinamidite-thioether ligands derived from Ugi's amine. These ferrocenyl P/S ligands have been employed for Pd-catalyzed asymmetric allylic alkylation

Table 3. The results of Pd-catalyzed asymmetric allylic amination using (*S,R_p*)-FerroNPS **2** as chiral ligand^a


Entry	Ligand	Concn (M)	Time (h)	Yield (%) ^b	ee (%) ^c
1	2a	0.25	3.5	98	89.1 (<i>S</i>)
2	2b	0.1 (THF)	48	86	91.5 (<i>S</i>)
3	2c	0.1	75	92	81.7 (<i>S</i>)

^a Reaction conditions: $[\text{Pd}(\eta^3\text{-C}_3\text{H}_5\text{Cl})_2]$ (2 mol %), ligand **2** (4.2 mol %), benzylamine (3.0 equiv), and EA as solvent at room temperature.

^b Isolated yield.

^c The % ee value was determined by HPLC on an OJ-H column (0.4 mL/min, *n*-Hex/*i*-PrOH = 85:15).

and amination, and excellent enantioselectivities and chemical yields were observed. Further investigation of other catalytic asymmetric reactions with these ferrocenyl N-P/S ligands is currently underway.

Acknowledgments

We thank the UGC Areas of Excellence Scheme (AoE/P-10/01) and The Hong Kong Polytechnic University (ASD) for financial support.

References

- (a) Trost, B. M.; Van Vranken, D. L. *Chem. Rev.* **1996**, *96*, 395–422; (b) Trost, B. M.; Crawley, M. L. *Chem. Rev.* **2003**, *103*, 2921–2943.
- Feringa, B. L. *Acc. Chem. Res.* **2000**, *33*, 346–353.
- Guiry, P. J.; Saunders, C. P. *Adv. Synth. Catal.* **2004**, *346*, 497–537.
- (a) Selvakumar, K.; Valentini, M.; Worle, M.; Pregosin, P. S.; Albinati, A. *Organometallics* **1999**, *18*, 1207–1215; (b) Pretot, R.; Pfaltz, A. *Angew. Chem., Int. Ed.* **1998**, *37*, 323–325.
- Gilbertson, S. R.; Lan, P. *Org. Lett.* **2001**, *3*, 2237–2240.
- (a) Albinati, A.; Pregosin, P. S.; Wick, K. *Organometallics* **1996**, *15*, 2419–2421; (b) Selvakumar, K.; Valentini, M.; Pregosin, P. S.; Albinati, A. *Organometallics* **1999**, *18*, 4591–4597.
- (a) Evans, D. A.; Campos, K. R.; Tedrow, J. S.; Michael, F. E.; Gagne, M. R. *J. Org. Chem.* **1999**, *64*, 2994–2995; (b) Evans, D. A.; Campos, K. R.; Tedrow, J. S.; Michael, F. E.; Gagne, M. R. *J. Am. Chem. Soc.* **2000**, *122*, 7905–7920.
- Mancheno, O. G.; Priego, J.; Cabrera, S.; Arrayas, R. G.; Llamas, T.; Carretero, J. C. *J. Org. Chem.* **2003**, *68*, 3679–3686.
- Cabreba, S.; Arrayas, R. G.; Carretero, J. C. *Angew. Chem., Int. Ed.* **2004**, *43*, 3944–3947.
- Mancheno, O. G.; Arrayas, R. G.; Carretero, J. C. *J. Am. Chem. Soc.* **2004**, *126*, 456–457.
- Cabreba, S.; Arrayas, R. G.; Carretero, J. C. *J. Am. Chem. Soc.* **2005**, *127*, 16394–16395.
- Cabreba, S.; Arrayas, R. G.; Alonso, I.; Carretero, J. C. *J. Am. Chem. Soc.* **2005**, *127*, 17938.
- (a) Colacot, T. J. *Chem. Rev.* **2003**, *103*, 3101–3118; (b) Dai, L.; Tu, T.; You, S.; Deng, W.; Hou, X. *Acc. Chem. Res.* **2003**, *36*, 659–667.
- Jia, X.; Li, X.; Lam, W. S.; Kok, Stanton H. L.; Xu, L.; Lu, G.; Yeung, C. H.; Chan, Albert S. C. *Tetrahedron: Asymmetry* **2004**, *15*, 2273–2278.
- Marquarding, D.; Klusacek, H.; Gokel, G.; Hoffmann, P.; Ugi, I. *J. Am. Chem. Soc.* **1970**, *92*, 5389–5393.
- Okoroafor, M. O.; Ward, D. L.; Brubaker, C. H. *Organometallics* **1988**, *7*, 1504–1511.
- Boaz, N. W.; Debenham, S. D.; Mackenzie, E. B.; Large, S. E. *Org. Lett.* **2002**, *4*, 2421–2424.